



Monongahela: Trihalomethanes (THMs)

Project Description :

Trihalomethanes (THMs), a family of halogenated disinfection byproducts, are ubiquitously present in finished drinking water. According to USEPA's Disinfectants/Disinfection Byproducts Rule (D/DBPR), the regulatory standard for total trihalomethanes (TTHMs) is 80 μ g/L, and water treatment plant operators are required to closely monitor TTHMs in finished water on a quarterly basis. Emergency funds from Colcom Foundation were used to fund sampling for THM at an elementary School near Fredericktown, Pennsylvania when TTHM levels exceeded this limit. This was a concern for the community since THMs can be linked to negative health conditions.

Although the relationship between intake bromide levels and TTHM in finished water is not clear, high bromide levels at the water authority's intake increase the risk of exceeding TTHM limits. WVWRI's routine monitoring as part of the 3RQ program has shown increasing bromide levels in the Monongahela River downstream of Masontown PA, reaching maximum levels at Brownsville and decreasing slightly downstream of the Youghiogheny River. High-bromide contributors in this portion of the Monongahela include: Whitely Creek; the Clyde Mine discharge at Clarksville, PA; and the Dillworth Mine discharge at Rice's Landing, PA.

Initial Study :

In 2016, with funding from the Colcom Foundation, 3RQ began collecting samples from the following water authorities: Charleroi, East Dunkard, PA American Brownsville, Southwest and TriCounty, along with 3RQ monitoring stations M61 and M82 (Figure 1). Samples were collected over a four week period during February 2016 and analyzed for TTHM and bromide levels.

The February sampling did not find any exceedances for TTHM at any of the sample locations (Figure 2). Additionally, bromide concentrations were low at all stations (non-detect shown as 1/2 the minimum detection limit (e.g. 0.010 mg/L). Our data showed no correlation between either bromide or chloride levels at the river intakes and TTHM. About 90% of the total THMs were either purely or dominantly chlorinated forms. This suggests that chlorination alone might be the dominant factor in raising THM levels during low river flow periods.

Sampling Locations:



Targeted

Study

Figure 1. Map of southwestern Pennsylvania showing the sampling locations.



Figure 2. TTHM concentrations at sampling locations in the Charleroi (C), East Dunkard (ED), PA American Brownsville (PAB), Tri County (TC) and Southwest (SW) systems (upstream of the intake and within the distribution system).

Follow-up Study :

A follow-up study was completed with funding from the United States Geological Survey (USGS). Samples were taken September 2018 through May 2019 from the Charleroi, TriCounty, and Southwest water authorities, which had all been part of the 2016 study.

In addition to the water samples, treatment related data was collected from the Borough of Charleroi Water Authority. This data included the daily chlorine dose added for the water treatment, quarterly organic carbon test in raw water, and quarterly THM check according to Disinfectants/Disinfection Byproducts Rule (D/DBPR) in the finished water. Additional analyses were conducted to investigate the controlling factors and remedies for TTHM formation in this water authority.

The objectives of this study were to:

- 1) Develop a statistical model for predicting TTHM levels using source water quality data.
- 2) Examine the effects of water treatment and distribution on TTHM formation.

Findings :

1) Total organic carbon (TOC) concentration in the intake water positively correlated with the TTHM formed in the finished water of distribution system. The correlation improved when plotted for individual treatment facility indicating that THM formation potential varies with the treatment practices of the individual facility.

2) The concentration of TOC was higher during the months of September and May when the temperature was also higher than other months. There was slight increase in TOC

Linear Regression Model for TTHM :

Based on the model, TTHM can be calculated from the following equation:

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TTHM (μg/L)= -2.95*chloride (mg/L) +
16.15*TOC (mg/L) + 4.65*Temp (°C)
-10.52
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The analysis showed that the developed model tends to under-estimate the TTHM concentration when TTHM concentrations are above 100 μ g/L. The inclusion of actual chlorine dose (other than chloride) and retention time may yield a better model to predict the TTHM in the finished water.

...with increase in temperature during March and April, but the TOC levels were still lower than the level in September. As a result, TTHM levels were found to be highest in each distribution point during September and May and reached above $80 \ \mu g/L$.

3) Our sampling showed that the bromide levels in the intake water were mostly below the reporting limit (0.01 mg/L). Thus, among the trihalomethanes, chloroform was the most dominant species in all the finished water samples.

Results :

There was a positive correlation of TTHM formation with the intake water total organic carbon (TOC) levels. Chloroform was the most dominant species in all the finished water samples, which can be seen in the figure below.







Figure 4. Comparison of measured TTHM vs the predicted TTHM Concentration from the model.

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